

One-Pot Transformation of Ph₂P(O)-Protected Ethynes: Deprotection Followed by Transition Metal-Catalyzed Coupling

Lifen Peng, Feng Xu, Yoshinori Suzuma, Akihiro Orita,* and Junzo Otera*

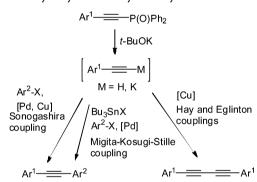
Department of Applied Chemistry, Okayama University of Science, Kita-ku, Okayama 700-0005, Japan

Supporting Information

ABSTRACT: Ph₂P(O)-protected ethynes were successfully transformed to arylethynes in one-pot manner through t-BuOKcatalyzed deprotection followed by Sonogashira coupling with aryl halide. The arylethynes were obtained similarly by Ph₂P(O)deprotection, stannylation of the resulting terminal ethynes, and Migita-Kosugi-Stille coupling. Deprotection followed by intramolecular Eglinton coupling could be carried out in one-pot to provide cyclic butadiynes.

he protection is one of the fundamental technologies in organic synthesis. 1 Although a number of protecting groups were developed and utilized in synthesis of structurally complicated compounds, new protecting groups are still being explored.² We have been involved in synthesis of acetylenic compounds having expanded π -systems and explored their applications to organic materials such as organic light-emitting diode (OLED),3 organic field-effect transistor (OFET),4 and dye-sensitized solar cell (DSSC).⁵ In synthesis of these arylethynes, Sonogashira coupling of aryl halide with ethynes protected with trialkylsilyl group such as trimethylsilyl (TMS) and t-buthyldimethylsilyl (TBDMS) and 2-hydroxy-2-propyl group is invoked routinely.6 However, this reaction, though powerful, frequently suffers from a severe drawback upon isolation of the product: the similar Rf values of starting compounds and products disturb isolation of the desired compound. In order to overcome this drawback, we developed a new protecting group, Ph₂P(O), which enabled easy isolation of the Sonogashira coupling product because of the high polarity, and exemplified the usefulness of this protection in the synthesis of phenyleneethynylenes. Herein, we have expanded the applicability of this protecting group to one-pot synthesis of arylethynes through deprotection followed by transition metalcatalyzed coupling reactions such as Sonogashira, Migita-Kosugi-Stille, Hay, and Eglinton couplings (Scheme 1).8 First, we tried deprotection of 1 followed by Sonogashira coupling of the resulting terminal ethyne with phenyl bromide in one-pot (Scheme 2). When a THF solution of 1 was treated with 1.2 equiv of t-BuOK at rt for 2 h, TLC analysis indicated that 1 was deprotected completely to give phenylethyne. After PhBr, Pd(PPh₃)₄, CuI, i-Pr₂NH, and toluene had been added to the THF solution, the mixture was heated at 80 °C for 20 h to give diphenylethyne in 72% yield. In deprotection of this protocol, t-BuOP(O)Ph2 was formed as a byproduct, but it did not

Scheme 1. One-Pot Transformation of Ph₂P(O)-Protected Ethyne to Arylethyne and Arylbutadiyne



disturb the following Sonogashira coupling. 10 Subjection of 2 to t-BuOK-catalyzed deprotection followed by Sonogashira coupling with 3 gave Ph₂P(O)-protected ethyne 4 in 72% yield. In this one-pot reaction, the addition of stoichiometric amount of t-BuOK enabled selective Ph₂P(O)-deprotection of 2 while the Ph₂P(O)-protection of 3 remained untouched. 11 Similarly, one-pot reaction of 5 with 6 provided 7 in 66% yield, and no decomposition of cyano group was observed in spite of the treatment of t-BuOK in the deprotection step. Ph₂P(O)protected ethyne 7 could be applied to one-pot Ph₂P(O)deprotection/Sonogashira coupling with 9-bromoanthracene to afford 8 in 76% yield.

We succeeded in synthesis of phenyleneethynylene having expanded π -system such as 8 by repeating one-pot deprotection/Sonogashira coupling protocol. The synthetic process

Received: October 6, 2013 Published: November 22, 2013

Scheme 2. One-Pot Synthesis of Arylethynes through Ph₂P(O)-Deprotection and Sonogashira Coupling

for phenyleneethynylene could be further compacted by subjection of crude product of the first one-pot protocol to the second (Scheme 3). When phosphorylethyne 2 was treated successively with t-BuOK and with aryl bromide 9 and Pd and Cu catalysts, phosphorylethyne 10 was provided, and subjection of the crude product 10 to the second deprotection/coupling with 4-(3,7-dimethyloctyloxy)phenyl iodide afforded 11 in 54% yield (based on 9). In this synthetic process, a filtration of the crude product 10 by a thin pad of silica gel was required. Otherwise, the second one-pot protocol was disturbed by the remaining oxidized transition metal catalyst(s) to provide 11 in a low yield. This compacted process of successive one-pot protocols could be applied to synthesis of 12: subjection of the phosphorylethyne 5 to deprotection/coupling protocols with 6 and 13 provided 12 in 43% yield.

The one-pot deprotection/coupling reaction protocol could be applied to Migata-Kosugi-Stille coupling ¹² as well as Sonogashira coupling. When phosphorylethyne **1** was subjected to deprotection (*t*-BuOK), stannylation (Bu₃SnOMe), and Migata-Kosugi-Stille coupling with iodide **14** (Pd₂(dba)₃ and *t*-Bu₃P), each step proceeded smoothly to give **15** in 96% yield (Scheme 4). ¹³ In this process, addition of 0.5 equiv of *t*-BuOK enabled the complete deprotection of the Ph₂P(O) group and the formation of stannylethyne **16** because MeOK which was produced by stannylation of the resulting potassium acetylide

with Bu₃SnOMe also served as a deprotection reagent.¹⁴ This deprotection/Migata-Kosugi-Stille coupling protocol proceeded smoothly in coupling between 1 and 17 to provide 18 in 77% yield. In this reaction, only stannylethyne 16 reacted with phenyl iodide moiety of 17, and terminal ethyne moiety of 17 remained untouched.¹⁵

We succeeded in synthesis of yne-diynes by taking advantage of Hay coupling 16 followed by one-pot deprotection/ Sonogashira coupling (Scheme 5). When a toluene solution of monophosphoryl-protected divne 19 and phenylethyne (20) was heated in the presence of CuCl, 21 was obtained in 76% yield. In this Hay coupling, homocoupling products 22 and 23 were produced as byproducts, but the high polarity of $Ph_2P(O)$ group enabled easy purification of the desired heterocoupling product 21 by column chromatography on silica gel (Rf = 0.55 for 21, 0.24 for 22, and 0.97 for 23 in AcOEt). Subjection of 21 to deprotection and Sonogashira coupling with 4-iodoanisole furnished vne-divne 24 in 74% yield. The similar Eglington coupling between terminal ethynes 25 and 26 followed by deprotection/Sonogashira coupling of the resulting butadiyne 27 with phenylbromide 28 provided nitro- and trifluoromethylsubstituted yne-diyne 29 ($60\% \times 62\%$ yield).

By invoking the Ph₂P(O)-assisted purification of the intermediate and copper-catalyzed butadiyne formation, we succeeded in synthesis of cyclic pentayne 30 (Scheme 6). Hay

Scheme 3. Compacted One-Pot Synthesis of Arylethynes through Ph₂P(O)-Deprotection and Sonogashira Coupling

Scheme 4. One-Pot Deprotection/Stannylation/Migata-Kosugi-Stille Coupling

coupling between monoprotected diyne **31** and iodoethyne **32** gave iodotriyne **33** in 76% yield, and Sonogashira coupling of the resulting iodide **33** with **31** provided bis-Ph₂P(O)-protected pentayne **34** in 78% yield. When **34** was subjected to *t*-BuOK-catalyzed deprotection followed by Cu(OAc)₂-catalyzed Eglington coupling,¹⁷ the expected cyclization

Scheme 5. Synthesis of Yne-diynes by Invoking Hay Coupling Followed by Deprotection/Sonogashira Coupling

Scheme 6. Synthesis of Cyclic Pentaynes by Invoking Hay, Sonogashira, and Eglinton Couplings

occurred to produce cyclic pentayne 30 in 55% yield. The final *in situ* deprotection/Eglington cyclization proceeded smoothly without isolation of deprotected-terminal ethyne 35. Although Haley has synthesized successfully the cyclic pentayne 30 by invoking the similar deprotection and cyclization of TMS-protected pentayne, 18 our protocol is more convenient to some extent than his process, which

requires transformation of $Ar-N_3Et_2$ to Ar-I by heating in a pressure bottle and *in situ* desilylation of TMS-proptected butadiyne/Sonogashira coupling of the terminal butadyne moiety with aryl iodide under strictly controlled reaction conditions.

In summary, we have established a new methodology of C-C bond formation by invoking in situ deprotection of Ph₂P(O) group/transition metal-catalyzed coupling of the resulting terminal ethyne. In one-pot deprotection/Sonogashira coupling protocol, unsymmetrically substituted aryleneethynylenes were obtained easily. When a stoichiometric amount of t-BuOK was used for deprotection, aryl halide having Ph₂P(O)-protected ethyne could be employed as a coupling counterpart, and the corresponding aryleneethynylene having Ph2P(O)-protected ethyne was obtained. By compaction of deprotection, stannylation and palladium-catalyzed coupling, one-pot deprotection/Migata-Kosugi-Stille coupling was realized. In this coupling protocol, in situ prepared stannylethyne reacted preferentially with aryl iodide, while unprotected terminal ethyne moiety of aryl iodide remained intact. Highly polar Ph₂P(O) protecting group enabled easy isolation of unsymmetrically substituted butadiynes which were obtained by copper-catalyzed oxidative coupling of teiminal ethynes, and the following one-pot deprotection/Sonogashira coupling afforded yne-diynes. In situ deprotection of Ph₂P(O) groups/ intramolecular Eglington coupling proceeded smoothly to give a cyclic pentayne.

EXPERIMENTAL SECTION

General. ¹H and ¹³C NMR spectra were recorded at 500 or 300 MHz and at 125 or 75 MHz, respectively, and calibrated with tetramethysilane (TMS) as an internal reference. High-resolution mass spectra (FAB) were recorded using *p*-nitrobenzyl alcohol as a matrix. Melting points (mp) were measured and uncorrected. All glassware was flame-dried prior to use, and all reactions were performed under nitrogen. Anhydrous solvents (THF, CH₂Cl₂) were purchased and used without further purification, and toluene and amines (Et₃N, *i*-Pr₂NH) were distilled from sodium and CaH₂, respectively, prior to use. Purification of the products was performed by flash column chromatography on silica gel (IR-60–63/210). Ethynes 17, ¹⁹ 26, ²⁰ and 32²¹ and 4-(3,7-dimethyloctyloxy)phenyl iodide²² were prepared according to the reported procedure.

Synthesis of Ph₂P(O)-Protected Ethynes 1, 2, and 5 (Representative Procedure for 1). A toluene solution (5.0 mL) of ethynylbenzene (109.8 μ L, 1.0 mmol), CuI (19.0 mg, 0.1 mmol), Ph₂PCl (220.6 μ L, 1.2 mmol), and Et₃N (277.2 μ L, 2.0 mmol) was stirred under nitrogen at 80 °C for 8 h. After usual workup with CH2Cl2 and NH4Claq, the combined organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was used for next step without purification. To a THF solution (10.0 mL) of the crude diphenyl-(phenylethynyl)phosphine was added H₂O₂aq (30%, 2.5 mL, 20.0 mmol) slowly at 0 °C, and the mixture was stirred at rt for 2 h. After workup with CH2Cl2 and water, the combined organic layer was washed with brine and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/AcOEt, 1:1) to afford diphenyl(phenylethynyl)phosphine oxide in a pure form (226.7 mg,

One (R = H):⁷ white powder; mp 94–96 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.39 (t, J = 7.6 Hz, 2H), 7.46 (d, J = 7.4 Hz, 1H), 7.48–7.52 (m, 4H), 7.56 (t, J = 7.3 Hz, 2H), 7.61 (d, J = 8.0 Hz, 2H), 7.89–7.93 (m, 4H).

Two (R = 4-MeO): white powder; mp 125–126 °C; ¹H NMR (500 MHz, CDCl₃): δ 3.84 (s, 3H), 6.89 (d, J = 8.8 Hz, 2H), 7.49–7.56 (m, 8H), 7.88–7.92 (m, 4H).

Five (R = CN):⁷ pale yellow powder; mp 163–165 °C; ¹H NMR (300 MHz, CDCl₃): δ 7.48–7.55 (m, 4H), 7.57–7.60 (m, 2H), 7.69 (br, 4H), 7.84–7.92 (m, 4H).

Synthesis of Ph₂P(O)-Protected Ethynes 3, 6, and 9 (Representative Procedure for 3). (i). Synthesis of Ethynyldiphenylphosphine Oxide. To a flask were added CuI (190.4 mg, 1.0 mmol), Ph₂PCl (1.8 mL, 10.0 mmol), trimethylsilylacetylene (1.7 mL, 12.0 mmol), triethylamine (2.8 mL, 20.0 mmol), and toluene (30.0 mL), and the mixture was stirred under nitrogen at 80 °C for 24 h. After workup with AcOEt/water, the organic layer was washed with aqueous NH₄Cl and brine, and dried over MgSO₄. After filtration, the solvents were evaporated. To the crude product were added THF (20.0 mL) and then 30% H_2O_2 aq (30%, 5.0 mL, 40.0 mmol) at 0 °C, and the mixture was stirred in the air at rt for 13 h. After workup with CH₂Cl₂/water, the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. To the crude product were added water (0.5 mL) and THF (50.0 mL), and then TBAF (1.0 M in THF, 1.0 mL, 1.0 mmol) at 0 °C, and the mixture was stirred in the air at rt for 5 h. After the solvents were evaporated, the crude product was subjected to column chromatography on silica gel (hexane/AcOEt, 1:1) to give ethynyldiphenylphosphine oxide (1.63 g, 72% yield in 3 steps) in a pure form.

Ethynyldiphenylphosphine oxide: white powder; H NMR (500 MHz, CDCl₃): δ 3.33 (d, J = 9.8 Hz, 1H), 7.48–7.52 (m, 4H), 7.56–7.59 (m, 2H), 7.83–7.87 (m, 4H); H C NMR (125 MHz, CDCl₃): δ 78.8 (d, J = 159.7 Hz), 94.0 (d, J = 27.4 Hz), 128.7 (d, J = 13.4 Hz), 130.9 (d, J = 11.4 Hz), 131.5, 132.5.

(ii). Sonogashira Coupling of 1-Bromo-3-lodobenzene with Ethynyldiphenylphosphine Oxide. A toluene solution (5.0 mL) of 1-bromo-3-iodobenzene (339.5 mg, 1.2 mmol), S1 (226.2 mg, 1.0 mmol), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol), CuI (9.5 mg, 0.05 mmol), and diisopropylamine (0.5 mL) was stirred under nitrogen at 80 °C for 15 h. After workup with CH₂Cl₂/NH₄Claq, the combined organic layer was washed with brine and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/AcOEt, 1:1) to afford 3 in a pure form (278.3 mg, 73% yield).

3: white powder; mp 98–99 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.26 (t, J = 7.9 Hz, 1H), 7.49–7.54 (m, 5H), 7.56–7.60 (m, 3H), 7.74 (s, 1H), 7.86–7.91 (m, 4H); ¹³C NMR (75 MHz, CDCl₃): δ 84.1 (d, J = 165.7 Hz), 103.1 (d, J = 29.2 Hz), 121.7 (d, J = 3.7 Hz), 122.2, 128.6 (d, J = 13.7 Hz), 130.0, 130.8 (d, J = 11.2 Hz), 132.3 (d, J = 3.1 Hz), 132.4 (d, J = 121.9 Hz), 133.7, 134.8, 134.9; ³¹P NMR (121 MHz, CDCl₃): δ 9.83; HRMS (FAB) calcd for $C_{20}H_{15}BrOP$ (M+H⁺): 381.0044, found 381.0049.

6:⁷ 46% yield; pale-yellow powder; mp 94–96 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.95 (t, J = 8.0 Hz, 1H), 7.49–7.53 (m, 4H), 7.56–7.59 (m, 3H), 7.79 (d, J = 7.9 Hz, 1H), 7.86–7.91 (m, 4H), 7.94 (s, 1H); ¹³C NMR (75 MHz, CDCl₃): δ 84.0 (d, J = 165.7 Hz), 93.4, 102.9 (d, J = 29.2 Hz), 121.6 (d, J = 4.0 Hz), 128.5 (d, J = 13.4 Hz), 129.9, 130.7 (d, J = 11.2 Hz), 131.4, 132.2 (d, J = 2.8 Hz), 132.4 (d, J = 121.9 Hz), 139.5, 140.5; ³¹P NMR (121 MHz, CDCl₃): δ 9.80.

9:⁷ 82% yield; white powder; mp 154–155 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.46 (d, J = 8.6 Hz, 2H), 7.49–7.59 (m, 8H), 7.86–7.91 (m, 4H); ¹³C NMR (75 MHz, CDCl₃): δ 83.7 (d, J = 166.3 Hz), 103.6 (d, J = 29.4 Hz), 118.2 (d, J = 4.3 Hz), 125.0, 128.3 (d, J = 13.7 Hz), 130.4 (d, J = 11.2 Hz), 131.5, 132.0 (d, J = 2.8 Hz), 132.2 (d, J = 121.9 Hz), 133.4 (d, J = 1.9 Hz); ³¹P NMR (121 MHz, CDCl₃): δ 9.88

Synthesis of Ph₂P(O)-Protected Ethynes 19, 25, and 31 (Representative Procedure for 19). A toluene solution (10 mL) of 1,3-diethynylbenzene (126.2 mg, 1.0 mmol), CuI (19.0 mg, 0.1 mmol), Ph₂PCl (220.6 μ L, 1.2 mmol), and Et₃N (277.2 μ L, 2.0 mmol) was stirred under nitrogen at 80 °C for 8 h. After workup with CH₂Cl₂/NH₄Claq, the combined organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was used for the next step without purification. To a THF solution (10.0 mL) of the crude product was added H₂O₂aq (30%, 2.5 mL, 20.0 mmol) at 0 °C, and the mixture was stirred at rt for 2 h. After workup with CH₂Cl₂/water, the combined organic layer was

washed with brine and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/AcOEt, 1:1) to afford **19** in a pure form (146.8 mg, 45% yield).

19: 7 white powder; mp 111–113 $^{\circ}$ C; 1 H NMR (500 MHz, CDCl₃): δ 3.12 (s, 1H), 7.35 (t, J = 8.0 Hz, 1H), 7.49–7.52 (m, 4H), 7.55–7.58 (m, 4H), 7.71 (s, 1H), 7.87–7.91 (m, 4H).

25: 48% yield; white powder; mp 145–147 °C; ¹H NMR (500 MHz, CDCl₃): δ 3.24 (s, 1H), 7.48–7.52 (m, 6H), 7.54–7.58 (m, 4H), 7.87–7.91 (m, 4H); ¹³C NMR (75.45 MHz, CDCl₃): δ 80.4, 82.6, 84.6 (d, J = 167.6 Hz), 104.4 (d, J = 29.4 Hz), 120.08, 120.13, 124.5, 128.7 (d, J = 13.7 Hz), 130.9 (d, J = 11.2 Hz), 132.2, 132.4, 132.7 (d, J = 121.9 Hz); ³¹P NMR (121 MHz, CDCl₃): δ 7.11; HRMS (FAB) calcd for $C_{22}H_{16}OP$ (M+H⁺): 327.0939, found 327.0946.

31: 45%; white powder; mp 121–122 °C; ¹H NMR (500 MHz, CDCl₃): δ 3.27 (s, 1H), 7.36 (t, J = 7.60 Hz, 1H), 7.41 (t, J = 7.65 Hz, 1H), 7.47–7.50 (m, 4H), 7.54–7.56 (m, 3H), 7.60 (d, J = 7.65 Hz, 1H), 7.94–7.99 (m, 4H); ¹³C NMR (75.45 MHz, CDCl₃): δ 81.3, 82.3, 86.4 (d, J = 167.8 Hz), 103.1 (d, J = 29.8 Hz), 123.1 (d, J = 3.7 Hz), 125.8 (d, J = 1.5 Hz), 128.5 (d, J = 13.6 Hz), 128.7, 130.2, 131.1 (d, J = 11.2 Hz), 132.2 (d, J = 3.1 Hz), 132.7, 132.8, 133.0 (d, J = 122.0 Hz); ³¹P NMR (121 MHz, CDCl₃): δ 6.65; HRMS (FAB) calcd for $C_{22}H_{16}$ OP (M+H⁺): 327.0939, found 327.0941.

One-Pot Ph₂P(O)-Deprotection of 1/Sonogashira Coupling with Phenyl Bromide (Representative). To a THF solution (10.0 mL) of 1 (302.3 mg, 1.0 mmol) was added t-BuOK (134.6 mg, 1.2 mmol). After the mixture had been stirred for 2 h under nitrogen at rt, bromobenzene (188.4 mg, 1.2 mmol), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol), CuI (9.5 mg, 0.05 mmol), toluene (16.0 mL), and diisopropylamine (0.5 mL) were added sequentially. The mixture was stirred under nitrogen at 80 °C for 15 h. After workup with CH₂Cl₂/NH₄Claq, the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane) to give diphenylacetylene in a pure form (128.3 mg, 72% yield).

Diphenylacetylene: ²³ white powder; mp 59–61 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.32–7.37 (m, 6H), 7.52–7.55 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 89.3, 123.2, 128.2, 128.3, 131.6.

4:⁷ white powder; mp 151–152 °C; ¹H NMR (500 MHz, CDCl₃): δ 3.84 (s, 3H), 6.89 (d, J = 8.8 Hz, 2H), 7.36 (t, J = 7.6 Hz, 1H), 7.47 (d, J = 8.8 Hz, 2H), 7.49–7.54 (m, 5H), 7.56–7.58 (m, 3H), 7.74 (s, 1H), 7.88–7.93 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 55.3 (d, J = 4.1 Hz), 83.3 (d, J = 167.8 Hz), 86.5, 90.9, 104.4 (d, J = 29.5 Hz), 113.99, 114.04, 114.6, 120.2 (d, J = 4.1 Hz), 124.4, 128.7 (d, J = 13.4 Hz), 130.9 (d, J = 11.3 Hz), 131.6, 132.3, 132.8 (d, J = 122.4 Hz), 133.1 (d, J = 4.2 Hz), 133.4, 135.1, 159.9; ³¹P NMR (121 MHz, CDCl₃): δ 9.78; HRMS (FAB) calcd for $C_{29}H_{22}O_{2}P$ (M+H⁺): 433.1357, found 433.1351.

7: white powder; mp 174–176 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.41 (t, J = 7.8 Hz, 1H), 7.49–7.53 (m, 4H), 7.56–7.61 (m, 6H), 7.65 (d, J = 8.5 Hz, 2H), 7.77 (s, 1H), 7.88–7.92 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 83.9 (d, J = 166.4 Hz), 88.9, 91.8, 103.8 (d, J = 29.4 Hz), 111.9, 118.3, 120.6 (d, J = 3.5 Hz), 123.0, 127.5, 128.7 (d, J = 13.4 Hz), 128.87, 128.94, 130.9 (d, J = 11.3 Hz), 132.1 (d, J = 6.8 Hz), 132.4, 132.7 (d, J = 121.9 Hz), 132.8, 133.7, 135.4; ³¹P NMR (121 MHz, CDCl₃): δ 6.82; HRMS (FAB) calcd for C₂₉H₁₉NOP (M+H⁺): 428.1204, found 428.1199.

8: yellow powder; mp 215–218 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.47 (t, J = 7.65 Hz, 1H), 7.54 (t, J = 7.05 Hz, 2H), 7.58 (d, J = 7.95 Hz, 1H), 7.60–7.64 (m, 2H), 7.64–7.68 (m, 4H), 7.78 (d, J = 7.65 Hz, 1H), 7.96 (s, 1H), 8.04 (d, J = 8.6 Hz, 2H), 8.47 (s, 1H), 8.64 (d, J = 8.6 Hz, 2H); ¹³C NMR (125 MHz, CDCl₃): δ 87.3, 88.4, 92.9, 99.5, 111.7, 116.7, 118.4, 122.8, 124.2, 125.7, 126.57, 126.62, 126.8, 127.9, 128.1, 128.2, 128.8, 131.2, 131.6, 132.08, 132.12, 132.7, 134.6 (d, J = 2.5 Hz); HRMS (FAB) calcd for $C_{31}H_{18}N$ (M+H+): 404.1439, found 404.1430.

One-Pot Me₃Si-Deprotection of Trimethyl(Phenylethynyl)-Silane/Sonogashira Coupling with Phenyl Bromide. To a THF solution (10.0 mL) of trimethyl(phenylethynyl)silane (174.3 mg, 1.0

mmol) was added t-BuOK (134.6 mg, 1.2 mmol) at 0 °C. After the mixture had been stirred for 2 h under nitrogen at 0 °C, bromobenzene (188.4 mg, 1.2 mmol), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol), CuI (9.5 mg, 0.05 mmol), toluene (16.0 mL), and diisopropylamine (0.5 mL) were added sequentially. The mixture was stirred under nitrogen at 80 °C for 15 h. After workup with CH₂Cl₂/NH₄Claq, the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane) to give diphenylacetylene in a pure form (83.8 mg, 47% yield).

Compacted One-Pot Synthesis of 11 from 2 through Ph₂P(O)-Deprotection/Sonogashira Coupling (Representative). To a THF solution (10.0 mL) of 2 (332.3 mg, 1.0 mmol) was added t-BuOK (134.6 mg, 1.2 mmol). After the mixture had been stirred for 2 h under nitrogen at rt, 9 (324.0 mg, 0.85 mmol), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol), CuI (9.5 mg, 0.05 mmol), toluene (16.0 mL), and diisopropylamine (0.5 mL) were added sequentially. The mixture was stirred under nitrogen at 80 °C for 15 h. After workup with CH₂Cl₂/NH₄Claq, the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was filtered by a thin pad of silica gel (hexane/AcOEt, 1:1), and the filtrate was concentrated. To a THF solution (10 mL) of the crude product 10 was added t-BuOK (134.6 mg, 1.2 mmol) at rt. After the mixture had been stirred for 2 h under nitrogen at rt, 4-(3,7dimethyloctyloxy)-1-iodobenzene (396.3 mg, 1.1 mmol), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol), CuI (9.5 mg, 0.05 mmol), toluene (16.0 mL), and diisopropylamine (0.5 mL) were added sequentially. The mixture was stirred under nitrogen at 80 °C for 15 h. After workup with CH₂Cl₂/NH₄Claq, the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/CH₂Cl₂, 5:1) to afford 11 in a pure form (213.3 mg, 54% yield, based on bromide).

11: white powder; mp 151–153 °C; ¹H NMR (500 MHz, CDCl₃): δ 0.87 (d, J = 6.75 Hz, 6H), 0.95 (d, J = 6.45 Hz, 3H), 1.15–1.35 (m, 6H), 1.51–1.62 (m, 2H), 1.67–1.68 (m, 1H), 1.80–1.86 (m, 1H), 3.83 (s, 3H), 3.97–4.05 (m, 2H), 6.86–6.89 (m, 4H), 7.45–7.48 (m, 8H); ¹³C NMR (75 MHz, CDCl₃): δ 19.6, 22.59, 22.63, 24.6, 28.0, 29.8, 36.1, 37.2, 39.2, 55.2, 66.4, 87.8, 87.9, 91.1, 91.3, 114.0, 114.6, 114.8, 115.2, 123.0, 123.1, 131.3, 133.0, 133.1, 159.3, 159.7; HRMS (FAB) calcd for C₃₃H₃₇O₂ (M+H⁺): 465.2794, found 465.2795.

12: white powder; mp 154–155 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.28–7.29 (m, 1H), 7.39 (t, J = 7.6 Hz, 1H), 7.54 (dd, J = 1.2 Hz, J = 8.0 Hz, 2H), 7.60–7.62 (m, 3H), 7.66 (d, J = 8.5 Hz, 2H), 7.71 (dt, J = 1.8 Hz, J = 7.9 Hz, 1H), 7.79 (s, 1H), 8.64 (d, J = 4.3 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃): δ 87.8, 88.3, 89.4, 92.6, 111.6, 118.4, 122.6, 122.8, 123.0, 127.2, 127.8, 128.6, 128.7, 132.0, 132.1, 132.4, 135.1, 136.2, 143.0, 150.1; HRMS (FAB) calcd for $C_{22}H_{13}N_2$ (M+H⁺): 305.1079, found 305.1086.

Synthesis of 15: One-Pot $Ph_2P(O)$ -Deprotection/Stannylation/Migata-Kosugi-Stille Coupling (Representative). To a THF solution (5.0 mL) of 1 (302.3 mg, 1.0 mmol) were added Bu_3SnOMe (353.2 mg, 316.8 μ L, 1.1 mmol) and t-BuOK (56.1 mg, 0.5 mmol) at rt, and the mixture was refluxed under nitrogen for 5 h. To the reaction mixture were added 14 (207.1 mg, 0.95 mmol), P(t-Bu)₃ (0.1 M in THF, 330.0 μ L, 0.033 mmol), and $Pd_2(dba)_3$ (13.7 mg, 0.015 mmol) at rt, and the mixture was stirred under nitrogen at rt for 4 h. After workup with diethyl ether (3 × 10.0 mL)/NH₄Faq (10%, 10.0 mL), the organic layer was dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane) to afford 15 in a pure form (175.3 mg, 96% yield, based on iodide).

15: ²⁴ colorless oil; ¹H NMR (500 MHz, CDCl₃): δ 2.37 (s, 3H), 7.16 (d, J = 7.9 Hz, 2H), 7.30–7.36 (m, 3H), 7.43 (d, J = 7.9 Hz, 2H), 7.52–7.54 (m, 2H); ¹³C NMR (75 MHz, CDCl₃): δ 21.5, 88.7, 89.5, 120.1, 123.4, 128.0, 128.3, 129.1, 131.4, 131.5, 138.3.

18: 25 white powder; mp 91–92 °C; ¹H NMR (500 MHz, CDCl₃): δ 3.18 (s, 1H), 7.35–7.36 (m, 3H), 7.46–7.50 (m, 4H), 7.52–7.54 (m,

2H); 13 C NMR (75 MHz, CDCl₃): δ 78.9, 83.2, 88.8, 91.3, 121.8, 122.8, 123.7, 128.4, 128.5, 131.4, 131.6, 132.0.

Synthesis of Yne-Diynes 24: Hay Coupling Followed by $Ph_2P(O)$ -Deprotection/Sonogashira Coupling (Representative). (i). Hay Coupling. A mixture of 19 (326.3 mg, 1.0 mmol), 20 (329.5 μ L, 3.0 mmol), CuCl (9.9 mg, 0.1 mmol), piperidine (50.0 μ L, 0.5 mmol), and toluene (10 mL) was stirred in the air at 75 °C for 15 h. After workup with CH_2Cl_2/NH_4Claq , the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/AcOEt, 1:1) to afford 21 (319.8 mg, 75% yield), 22 (55.3 mg, 17%), and 23 (215.4 mg, 71% (based on 20)) in pure forms.

21: pale-yellow powder; mp 166–167 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.33–7.41 (m, 4H), 7.49–7.54 (m, 6H), 7.56–7.59 (m, 4H), 7.74 (s, 1H), 7.87–7.92 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 73.4, 75.3, 79.6, 82.4, 83.8 (d, J = 167.0 Hz), 103.7 (d, J = 29.4 Hz), 120.5 (d, J = 4.1 Hz), 121.4, 122.7, 128.4 (d, J = 5.2 Hz), 128.7 (d, J = 13.4 Hz), 128.9 (d, J = 8.8 Hz), 129.4 (d, J = 3.1 Hz), 130.9 (d, J = 11.3 Hz), 132.4, 132.5, 132.7 (d, J = 121.9 Hz), 132.8, 134.3, 136.1; ³¹P NMR (121 MHz, CDCl₃): δ 6.36; HRMS (FAB) calcd for $C_{30}H_{20}$ OP (M+H*): 427.1252, found 427.1259.

22: pale-yellow powder; mp 198–201 °C; ¹H NMR (300 MHz, CDCl₃): δ 7.38 (t, J = 7.8 Hz, 2H), 7.48–7.61 (m, 16H), 7.75 (s, 2H), 7.86–7.94 (m, 8H); ¹³C NMR (75 MHz, CDCl₃): δ 74.8, 80.4, 83.9 (d, J = 166.0 Hz), 103.6 (d, J = 29.2 Hz), 120.6 (d, J = 4.0 Hz), 122.3, 128.7 (d, J = 13.4 Hz), 128.9, 130.9 (d, J = 11.5 Hz), 132.4 (d, J = 2.8 Hz), 132.6 (d, J = 122.2 Hz), 133.0, 134.4, 136.1; ³¹P NMR (121 MHz, CDCl₃): δ 6.90; HRMS (FAB) calcd for $C_{44}H_{29}O_2P_2$ (M+H⁺): 651.1643, found 651.1641.

23: 26 white powder; mp 83–84 °C; 1 H NMR (500 MHz, CDCl₃): δ 7.32–7.39 (m, 6H), 7.52–7.54 (m, 4H); 13 C NMR (500 MHz, CDCl₃): δ 73.9, 81.5, 121.8, 128.4, 129.2, 132.5.

(ii). $Ph_2P(O)$ -Deprotection/Sonogashira Coupling. To a THF solution (10.0 mL) of **21** (213.2 mg, 0.5 mmol) was added t-BuOK (67.3 mg, 0.6 mmol) at rt. After the mixture had been stirred under nitrogen at rt for 2 h, 1-iodo-4-methoxybenzene (128.7 mg, 0.55 mmol), Pd(PPh₃)₄ (28.9 mg, 0.025 mmol), CuI (4.8 mg, 0.025 mmol), toluene (16.0 mL), and diisopropylamine (0.25 mL) were added sequentially. The mixture was stirred under nitrogen at 80 °C for 15 h. After workup with CH_2Cl_2/NH_4Claq , the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/ CH_2Cl_2 , 8:1) to give **24** in a pure form (123.0 mg, 74% yield).

24: white powder; mp 126–128 °C; ¹H NMR (500 MHz, CDCl₃): δ 3.84 (s, 3H), 6.89 (d, J = 8.8 Hz, 2H), 7.29–7.40 (m, 4H), 7.45–7.50 (m, 4H), 7.54 (d, J = 6.7 Hz, 2H), 7.67 (s, 1H); ¹³C NMR (125 MHz, CDCl₃): δ 55.3 (d, J = 4.1 Hz), 73.8, 74.4, 80.7, 81.8, 86.9, 90.4, 114.0, 114.1, 114.9, 121.7, 122.1, 124.2, 128.5, 129.3, 131.7, 132.0, 132.5, 133.2 (d, J = 8.3 Hz), 135.2 (d, J = 3.1 Hz), 159.8; HRMS (FAB) calcd for $C_{25}H_{17}O$ (M+H⁺): 333.1279, found 333.1285.

27: white powder; mp 253–255 °C; 1 H NMR (500 MHz, CDCl₃): δ 7.49–7.54 (m, 6H), 7.56–7.59 (m, 4H), 7.60–7.64 (m, 4H), 7.87–7.91 (m, 4H); 13 C NMR (125 MHz, CDCl₃): δ 75.7, 76.4, 81.38, 81.44, 85.4 (d, $J_{\rm C-P}$ = 166.4 Hz), 104.1 (d, $J_{\rm C-P}$ = 29.4 Hz), 120.7, 120.8, 123.6 (q, $J_{\rm C-F}$ = 272.0 Hz), 123.8, 125.2 (d, $J_{\rm C-P}$ = 1.5 Hz), 125.4 (d, $J_{\rm C-F}$ = 3.1 Hz), 128.7 (d, $J_{\rm C-P}$ = 13.4 Hz), 130.95 (d, $J_{\rm C-P}$ = 10.8 Hz), 130.98 (q, $J_{\rm C-F}$ = 32.5 Hz), 132.4, 132.5 (d, $J_{\rm C-P}$ = 8.9 Hz), 132.6 (d, $J_{\rm C-P}$ = 122.0 Hz), 132.8; 19 F NMR (282 MHz, CDCl₃): δ –94.03; 31 P NMR (121 MHz, CDCl₃): δ 6.51; HRMS (FAB) calcd for $C_{31}H_{19}F_{3}$ OP (M+H $^{+}$): 495.1126, found 495.1122.

29: white powder; mp 159–161 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.52–7.57 (m, SH), 7.62 (q, J = 8.6 Hz, 4H), 7.83 (d, J = 7.9 Hz, 1H), 8.19–8.21 (m, 1H), 8.38 (s, 1H); ¹³C NMR (125 MHz, CDCl₃): δ 75.6, 76.0, 80.9, 82.0, 89.5, 91.1, 121.9, 123.2, 123.3, 123.7 (q, J_{C-F} = 272.1 Hz), 124.6, 125.4 (q, J_{C-F} = 2.9 Hz), 126.4, 126.5, 129.4, 130.9 (q, J_{C-F} = 32.8 Hz), 131.8, 132.6, 132.7, 137.2, 148.2; ¹°F NMR (282 MHz, CDCl₃): δ –63.49; HRMS (FAB) calcd for $C_{25}H_{13}F_3NO_2$ (M +H*): 416.0898, found 416.0902.

Synthesis of Cyclic Pentayne 30. (i). Synthesis of 33. A mixture of 31 (326.3 mg, 1.0 mmol), 32 (912.1 mg, 4.0 mmol), CuCl (9.9 mg, 0.1 mmol), piperidine (50.0 μ L, 0.5 mmol), and toluene (10.0 mL) was stirred in the air at 75 °C for 15 h. After workup with CH₂Cl₂/NH₄Claq, the organic layer was washed with brine, and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/AcOEt, 1:1) to afford 33 in a pure form (419.8 mg, 76% yield).

33: yellow powder; mp 129–131 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.07 (dt, J = 1.2 Hz, J = 7.6 Hz, 1H) 7.32 (t, J = 7.6 Hz, 1H), 7.36–7.41 (m, 1H), 7.42–7.44 (m, 2H), 7.48–7.50 (m, 6H), 7.60–7.62 (m, 2H), 7.85 (d, J = 7.9 Hz 1H), 8.00–8.04 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 76.9, 78.5, 80.7, 84.6, 86.9 (d, J = 166.5 Hz), 100.8, 102.6 (d, J = 29.4 Hz), 123.6 (d, J = 4.1 Hz), 125.2 (d, J = 2.1 Hz), 127.7, 128.2, 128.7 (d, J = 12.4 Hz), 129.2, 130.3, 130.4, 130.5, 131.0 (d, J = 11.4 Hz), 132.1, 132.8 (d, J = 122.0 Hz), 133.0 (d, J = 6.6 Hz), 133.9, 138.9; ³¹P NMR (121 MHz, CDCl₃): δ 6.60; HRMS (FAB) calcd for $C_{30}H_{19}IOP$ (M+H $^+$): 553.0218, found 553.0212.

(ii). Synthesis of 34. A toluene solution (10.0 mL) of 33 (276.2 mg, 0.5 mmol), 31 (195.9 mg, 0.6 mmol), Pd(PPh₃)₄ (28.9 mg, 0.025 mmol), CuI (4.8 mg, 0.025 mmol), and diisopropylamine (0.25 mL) was stirred under nitrogen at rt for 28 h. After workup with CH₂Cl₂/NH₄Claq, the combined organic layer was washed with brine and dried over MgSO₄. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (AcOEt) and recrystallization from THF/hexane to afford 34 in a pure form (292.8 mg, 78% yield).

34: pale-yellow powder; mp 114–116 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.21–7.24 (m, 2H), 7.30–7.34 (m, 3H), 7.35–7.39 (m, 5H), 7.40–7.48 (m, 11H), 7.59–7.63 (m, 3H), 7.92–7.98 (m, 8H); ¹³C NMR (125 MHz, CDCl₃): δ 77.6, 78.6, 80.4, 81.8, 86.4 (d, J = 167.9 Hz), 86.9 (d, J = 166.5 Hz), 91.9, 92.4, 102.6 (d, J = 29.4 Hz), 103.5 (d, J = 30.0 Hz), 122.1 (d, J = 3.6 Hz), 123.4 (d, J = 4.1 Hz), 123.9, 125.3 (d, J = 2.0 Hz), 126.2, 126.4 (d, J = 2.0 Hz), 128.4 (d, J = 12.4 Hz), 128.5 (d, J = 11.9 Hz), 128.59, 128.61, 128.7, 129.1, 129.2, 130.3, 130.5, 130.90 (d, J = 11.3 Hz), 130.91 (d, J = 11.4 Hz), 132.1, 132.4, 132.6, 132.7 (d, J = 122.0 Hz), 132.9, 133.0 (d, J = 122.0 Hz), 133.1, 133.2; ³¹P NMR (121 MHz, CDCl₃): δ 7.07; HRMS (FAB) calcd for $C_{52}H_{33}O_{7}P_{2}$ (M+H+): 751.1956, found 751.1957.

(iii). Synthesis of 30. To a THF solution (4.0 mL) of 34 (75.1 mg, 0.1 mmol) was added t-BuOK (26.9 mg, 0.24 mmol) at rt, and the mixture was stirred under nitrogen at rt for 4 h. To the reaction mixture were added pyridine (20.0 mL) and methanol (20.0 mL), and the mixture was added to $Cu(OAc)_2(H_2O)$ (399.3 mg, 2.0 mmol) in a mixture of pyridine (18.5 mL), methanol (18.5 mL), and Et_2O (3.1 mL) at 46 °C by a syringe over 7 h. After the mixture had been stirred in the air overnight, the mixture was concentrated. After addition of CH_2Cl_2 and 10% HCl, the mixture was stirred vigorously for 30 min. The CH_2Cl_2 layer was separated, washed with Na_2CO_3aq solution, water, and brine, and dried over $MgSO_4$. After filtration, the solvents were evaporated. The crude product was subjected to column chromatography on silica gel (hexane/ CH_2Cl_2 , 8:1) to afford 30 in a pure form (19.2 mg, 55% yield).

30: ¹⁸ pale-yellow powder; mp 159 °C (decomp); ¹H NMR (500 MHz, CDCl₃): δ 7.21–7.29 (m, 6H), 7.32–7.35 (m, 2H), 7.40 (d, J = 7.0 Hz, 2H), 7.47 (d, J = 7.6 Hz, 2H); ¹³C NMR (125 MHz, CDCl₃): δ 77.7, 80.8, 81.6, 82.2, 93.0, 123.2, 126.9, 128.1, 128.5, 129.0, 129.1, 130.5, 133.7, 134.5.

■ ASSOCIATED CONTENT

Supporting Information

¹H, ¹³C, ¹⁹F, and ³¹P NMR spectra for all products. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Authors

*E-mail: orita@high.ous.ac.jp. *E-mail: otera@high.ous.ac.jp.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This work was supported by a Grant-in-Aid for Scientific Research on Innovative Areas "Organic Synthesis based on Reaction Integration. Development of New Methods and Creation of New Substances" (No. 2105), matching fund subsidy for private universities from the Ministry of Education, Culture, Sports, Science and Technology, Japan, the Japan Society for the Promotion of Science (JSPS) through its "Funding Program for World-Leading Innovative R&D on Science and Technology (FIRST Program)" and Okayama Prefecture Industrial Promotion Foundation.

REFERENCES

- (1) Greene, T. W.; Wuts, P. G. M. Protective Groups in Organic Synthesis, 3th ed.; John Wiley & Sons, Inc.: New York, 1999.
- (2) For recent examples, see: (a) For hydroxyl group of carbohydrate, Rankin, G. M.; Maxwell-Cameron, I.; Painter, G. F.; Larsen, D. S. J. Org. Chem. 2013, 78, 5264. (b) For amino group of carbazole, Urones, B.; Gómez Arrayás, R.; Carretero, J. C. Org. Lett. 2013, 15, 1120. (c) For vic-diol, Balbuena, P.; Gonçalves-Pereira, R.; Jiménez Blanco, J. L.; García-Moreno, M. I.; Lesur, D.; Ortiz Mellet, C.; García Fernández, J. M. J. Org. Chem. 2013, 78, 1390. (d) For carboxamide, Muranaka, K.; Ichikawa, S.; Matsuda, A. J. Org. Chem. 2011, 76, 9278. (e) For boronic acid, Ihara, H.; Koyanagi, M.; Suginome, M. Org. Lett. 2011, 13, 2662. (f) For alcohol and amine, Liang, H.; Corey, E. J. Org. Lett. 2011, 13, 4120.
- (3) (a) Mao, G.; Orita, A.; Fenenko, L.; Yahiro, M.; Adachi, C.; Otera, J. *Mater. Chem. Phys.* **2009**, *115*, 378. (b) Fenenko, L.; Shao, G.; Orita, A.; Yahiro, M.; Otera, J.; Svechnikov, S.; Adachi, C. *Chem. Commun.* **2007**, 2278.
- (4) (a) Matsuo, D.; Yang, X.; Hamada, A.; Morimoto, K.; Kato, T.; Yahiro, M.; Adachi, C.; Orita, A.; Otera, J. Chem. Lett. 2010, 39, 1300. (b) Oyamada, T.; Shao, G.; Uchiuzou, H.; Nakanotani, H.; Orita, A.; Otera, J.; Yahiro, M.; Adachi, C. Jpn. J. Appl. Phys. Part II 2006, 45, L1331.
- (5) (a) Yang, X.; Kajiyama, S.; Fang, J.-K.; Xu, F.; Uemura, Y.; Koumura, N.; Hara, K.; Orita, A.; Otera, J. *Bull. Chem. Soc. Jpn.* **2012**, 85, 687. (b) Yang, X.; Fang, J.-K.; Suzuma, Y.; Xu, F.; Orita, A.; Otera, J.; Kajiyama, S.; Koumura, N.; Hara, K. *Chem. Lett.* **2011**, 40, 620.
- (6) (a) Sonogashira, K.; Tohda, Y.; Hagihara, N. Tetrahedron Lett. 1975, 4467. (b) Tohda, Y.; Sonogashira, K.; Hagihara, N. Synthesis 1977, 777. (c) Takahashi, S.; Kuroyama, Y.; Sonogashira, K.; Hagihara, N. Synthesis 1980, 627. For a recent review: (d) Chinchilla, R.; Nájera, C. Chem. Rev. 2007, 107, 874.
- (7) Yang, X.; Matsuo, D.; Suzuma, Y.; Fang, J.-K.; Xu, F.; Orita, A.; Otera, J.; Kajiyama, S.; Koumura, N.; Hara, K. Synlett 2011, 2402.
- (8) For reviews of space-integrated one-pot reactions: (a) Yoshida, J.; Saito, K.; Nokami, T.; Nagaki, A. *Synlett* **2011**, 1189. (b) Suga, S.; Yamada, D.; Yoshida, J. *Chem. Lett.* **2010**, 39, 404.
- (9) When Me₃Si-protected phenylethyne was used for this one-pot deprotection/Sonogashira coupling instead of 1, diphenylethyne was obtained in 47% yield. See Experimental Section.
- (10) This dephosphorylation proceeded through C(sp)-P bond cleavage. For a review of transition-metal-mediated C-P bond cleavages of tertiary phosphines: Garrou, P. E. Chem. Rev. 1985, 85, 171. For lithium-induced reductive $C(sp^2)$ -P bond cleavage of phoshine: Dogan, J.; Schulte, J. B.; Swiegers, G. F.; Wild, S. B. J. Org. Chem. 2000, 65, 951. For $C(sp^2)$ -P bond cleavage of phosphine

- oxide by a yttrium complex: Lu, E.; Chen, Y.; Zhou, J.; Leng, X. Organometallics 2012, 31, 4574.
- (11) For complete deprotection of 2, 1.2 equiv of t-BuOK were used, although 3 would undergo also deprotection by an excess amount of t-BuOK. In order to consume phosphoryl bromide 3 completely in the coupling stage, 0.85 equiv of 3 was used for the Sonogashira coupling. Otherwise the remaining phosphoryl-ethyne 2 and -bromide 3 would prevent easy isolation of 4 because of similar Rf values of 2, 3, and 4 (Rf = 0.30 for 2, 0.32 for 3, and 0.29 for 4, hexane/AcOEt (1:1)).
- (12) (a) Kosugi, M.; Fugami, K. Handbook of Organopalladium Chemistry for Organic Synthesis, Negishi, E., Ed.; Wiley: New York, 2002. (b) Tsuji, J. Palladium Reagents and Catalysts; Wiley: Chichester, U.K., 2004. (c) Stille, J. K. Angew. Chem., Int. Ed. Engl. 1986, 25, 508. (d) Milstein, D.; Stille, I. K. J. Am. Chem. Soc. 1978, 100, 363.
- (13) It was reported that Me₃Si-CCR was transformed successfully to Bu₃Sn-CCR by treatment with (Bu₃Sn)₂O in the presence of a catalytic amount of tetrabutylammonium fluoride (TBAF). Warner, B. P.; Buchwald, S. L. *J. Org. Chem.* **1994**, *59*, 5822.
- (14) When 0.3 equiv of t-BuOK and 1.1 equiv of Bu₃SnOMe were added for dephosphorylation/stannylation, TLC analysis indicated a poor yield (<50%) of **16** after reflux for 12 h. In contrast to this, when 0.5 or 1.0 equiv of t-BuOK was used, complete transformation of **1** to **16** was observed on TLC analysis.
- (15) Kiyokawa, K.; Tachikake, N.; Yasuda, M.; Baba, A. Angew. Chem., Int. Ed. 2011, 50, 10393.
- (16) For high polarity-assisted syntheses of diynes, see: (a) For (3-cyanopropyl)dimethylsilyl (CPDMS) protection, Mössinger, D.; Jester, S.-S.; Sigmund, E.; Müller, U.; Höger, S. *Macromolecules* **2009**, *42*, 7974. (b) For sulfinyl carboxylate, Nicolaou, K. C.; Zipkin, R. E.; Petasis, N. A. *J. Am. Chem. Soc.* **1982**, *104*, 5558.
- (17) For desilylation and Cu(OAc)₂-catalyzed coupling for synthesis of diynes, see: Wan, W. B.; Brand, S. C.; Pak, J. J.; Haley, M. M. Chem.—Eur. J. **2000**, *6*, 2044.
- (18) Haley, M. M.; Bell, M. L.; English, J. J.; Johnson, C. A.; Weakley, T. J. R. J. Am. Chem. Soc. 1997, 119, 2956.
- (19) Goeb, S.; Ziessel, R. Org. Lett. 2007, 9, 737.
- (20) Alameddine, B.; Aebischer, O. F.; Amrein, W.; Donnio, B.; Deschenaux, R. Chem. Mater. 2005, 17, 4798.
- (21) Hashmi, A. S. K.; Ingo Braun, M. W.; Rudolph, M.; Rominger, F. Angew. Chem., Int. Ed. 2012, 51, 10633.
- (22) Aparicio, F.; García, F.; Fernández, G.; Matesanz, E.; Sánchez, L. Chem.—Eur. J. 2011, 17, 2769.
- (23) Mio, M. J.; Kopel, L. C.; Braun, J. B.; Gadzikwa, T. L.; Hull, K. L.; Brisbois, R. G.; Markworth, C. J.; Grieco, P. A. *Org. Lett.* **2002**, *4*, 3199.
- (24) Peña-López, M.; Ayán-Varela, M.; Sarandeses, L. A.; Sestelo, J. P. Chem.—Eur. J. **2010**, *16*, 9905.
- (25) Li, J.; Huang, P. C. Beilstein J. Org. Chem. 2011, 7, 426.
- (26) Gelderen, L.; Rothenberg, G.; Calderone, V. R.; Wilson, K.; Shiju, N. R. Appl. Organometal. Chem. 2013, 27, 23.